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Chem. 121, Sect 012, Quiz 3

Fall, 2012, 50 points

1. Assign the absolute configuration (R or S) to each of the chirality centers in the following molecules. (10 pts)

(b) 
$$H \xrightarrow{CH} CH_3$$
  $CH_3$ 

2. Give the product(s) formed in the following reaction. (10 pts)

3. For the following molecules indicate which would give a positive test with (a) potassium permanganate (b)  $KI/I_2$  (iodoform test). There may be more than one correct answer. (10 pts)

(a) 
$$CH_2CH_3$$
 (b)  $CH_3CH_2$  (c)  $CH_3-C-CH_2CH_3$  (d)  $CH_3CH_2-C-CH_2CH_3$   $CH_2CH_3$ 

- 4. In the preparation of 1-bromobutane from 1-butanol, aqueous sulfuric acid and sodium bromide (a) show the complete reaction and the complete reaction mechanism. (b) Explain the purpose of the gas-trap? (c) In the work-up, why did we wash the organic layer once with 80% sulfuric acid? (10 pts)
- 5. In the preparation of cyclohexanone from cyclohexanol using sodium hypochlorite (NaOCl, household bleach) and acetic acid (CH<sub>3</sub>CO <sub>2</sub>H) (a) show the overall reaction (no mechanism required). (b) In the work-up we added sodium bicarbonate (Na<sub>2</sub>CO<sub>3</sub>) to the initial distillate. Why did we do this? Show any reaction that may have occurred. (c) Why did we add the methylene chloride (CH<sub>2</sub>Cl<sub>2</sub>)? Explain briefly. (10 pts)