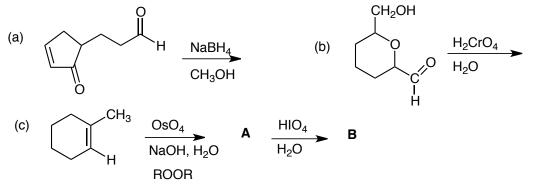
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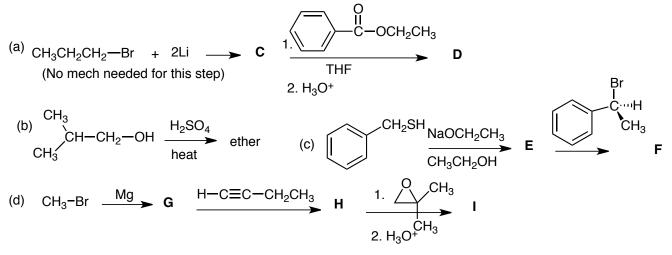
## Long Island University, Department of Chemistry Exam 1, 100 pts, Spring, 2012

1. Identify the following molecules: (a) formula ( $C_4H_8O_2$ ). IR: 1745 cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$  1.1, triplet, 3H; 2.1, singlet, 3H; 3.8, quartet, 2H. (b) formula ( $C_{12}H_{15}O_2$ ). IR: 3200 – 3500 and 1723 cm<sup>-1</sup>. <sup>1</sup>H NMR:  $\delta$ ; 1.0, doublet, 6H; 2.1, septet, 1H; 2.2, triplet, 2H; 2.4, triplet, 2H; 7.2, doublet, 2H; 7.4, doublet, 2H. (20 pts)

2. Give the product of the following reactions. It is not necessary to show the full mechanism. Be sure to show the stereochemistry where appropriate. (20 pts)



3. For the following reactions, give the product(s) and the complete reaction mechanisms by which they are formed. (40 pts)



4. Synthesize two of the following three molecules as shown. For extra credit do all three. (20 pts)

