## Long Island University, Department of Chemistry

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Exam 1, 100 pts, Spring, 2011

1. Identify the following molecule: formula ( $C_8H_{14}O_4$ ); IR: 1745 cm<sup>-1</sup>; <sup>1</sup>H NMR:  $\delta$  1.2, 6H, triplet; 2.2, singlet, 4H, 3.8, quartet, 4H. (10 pts)

2. Give the product of the following reactions. It is not necessary to show the full mechanism. (30 pts)

(a) 
$$OCH_3$$
  $OCH_3$   $OCH_4$   $OCH_4$   $OCH_5$   $OCH_5$ 

3. For the following reactions, give the product(s) and the complete reactions mechanisms by which they are formed. Pay careful attention to stereochemistry where appropriate. (30 pts)

(a) OH + OH 
$$\frac{H_2SO_4}{\text{heat}}$$
 3 products (b)  $\frac{CH_3CH_2O^-Na^+}{CH_3CH_2OH}$  (c)  $\frac{2 \text{ Li}}{\text{THF}}$  A  $\frac{O}{CH_3}$  B

4. Show how the following transformation occurs, giving all the steps of the mechanism including all intermediates. (10 pts)

5. Synthesize **two** of the following three molecules as shown. For extra credit do all three. (20 pts)

(a) 
$$H-C \equiv C-H$$
  $\longrightarrow$   $CH_3CH_2$   $CHCH_2CH_3$  (b)  $H-C \equiv C-H$   $\longrightarrow$   $CH_2CH_3$   $CH_2-CH-CH_2CH_3$   $CH_2-CH-CH_2CH_3$